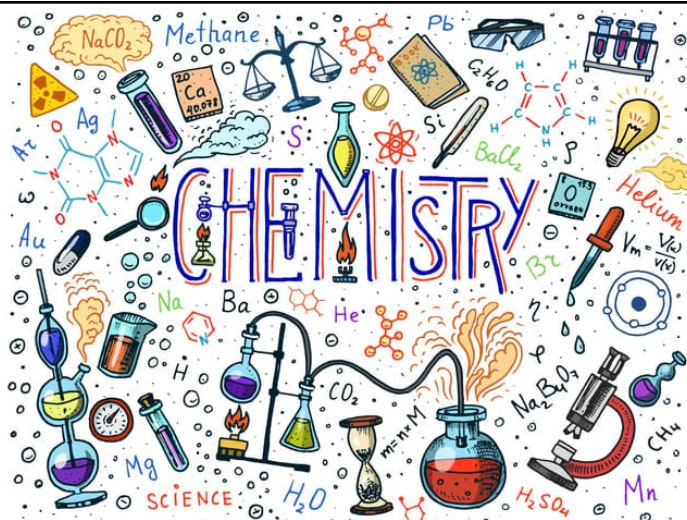


*BHHS Chemistry*  
**Chemical Concepts with Lab**  
2023-24

**Instructor Information**

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<b>Description and Credits</b>	Chemistry is a survey of chemical concepts, with a focus on practical applications of chemistry and the impact of chemistry on the environment, society, economy and individuals.
<b>Textbook</b>	Chemistry by Prentice Hall (2008), ISBN 0-13-251210-6 Openstax Chemistry 2e, free online. <a href="https://openstax.org/details/books/chemistry-2e">https://openstax.org/details/books/chemistry-2e</a>
<b>Required Materials</b>	Binder, Paper, Folder, Pencils, Calculator, Highlighters, Safety goggles* *Fully protective chemical splash goggles will be provided. However, if you would like your own new pair or prefer a more comfortable model you may wish to purchase your own to use. Vented, anti-fog safety goggles may be found through Amazon or other companies.

**COURSE EXPECTATIONS**

RESPECT

- Respect your peers: Consider your words and actions carefully- keep them kind.
- Respect your teacher: I'm here to help you. Communicate with me if an issue arises.
- Respect your environment: Practice ALL safety rules in the classroom and keep it tidy.
- Respect yourself: Do not settle for less- only you can push yourself all the way to success.

RESPONSIBILITY

- Be on time to class. Be prepared with all materials.
- Complete and turn in work on time. If you are absent, check google classroom for assignments.
- Communicate with your teacher if you are experiencing an issue that prevents you from completing assignments.
- Own up to your mistakes and FIX THEM. Ask for help when you need it. Act maturely.

PRESENCE

- Show up: you cannot learn if you are not here. Be not only present in body, but present in mind.
- Participate in class activities, as questions so that you may understand. Your presence will greatly affect your learning and therefore have a large impact on your grade.

HONESTY

- DO NOT CHEAT!! If you are not doing your own work, it is doing you no good.
- Cheating and Plagiarism are violations of BHHS & SPSCC's policy and may result in disciplinary action and/or an F for the quiz, lab or assessment.

Grading Scale (%)		Course Topics	Learning Outcomes
A	93-100	<ul style="list-style-type: none"> <li>● Systems of measurement and unit conversion               <ul style="list-style-type: none"> <li>○ Metric system</li> <li>○ Scientific notation</li> </ul> </li> <li>● Properties and transformations of matter               <ul style="list-style-type: none"> <li>○ Chemical and physical properties</li> <li>○ Chemical and physical change</li> </ul> </li> <li>● Structure of matter               <ul style="list-style-type: none"> <li>○ Periodic table</li> <li>○ Atomic theory</li> <li>○ Conservation of matter</li> <li>○ Elements, compounds, and mixtures</li> <li>○ Solids, liquids, and gasses</li> </ul> </li> <li>● Symbolic language of chemistry               <ul style="list-style-type: none"> <li>○ Element symbols</li> <li>○ Compound formulas and names</li> </ul> </li> <li>● Chemical reaction equations</li> <li>● Energy sources and uses</li> </ul>	<ul style="list-style-type: none"> <li>● Identify the role of chemical reactions in daily life</li> <li>● Describe visible and molecular characteristics of solids, liquid, and gases</li> <li>● Compare the characteristics of pure substances versus mixtures</li> <li>● Interpret chemical reactions using correct chemical notation and vocabulary</li> <li>● Explain experimental results and observations in terms of physical and chemical phenomena</li> <li>● Apply simple mathematical reasoning to chemical systems</li> </ul>
A-	90-92		
B+	87-89		
B	83-86		
B-	80-82		
C+	77-79		
C	73-76		
C-	70-72		
D+	67-69		
D	60-66		
F	Below 60		

Method of Evaluation and Class Policies	
<b>Assessments</b> <b>60.0%</b>	<ul style="list-style-type: none"> <li>● Students will be tested on each unit in Chemistry, and may use their card stock note sheets, periodic table and charts.</li> <li>● Students may select to complete test corrections or a retest. All assignments must be completed in order to complete corrections or retakes.</li> </ul>
<b>Labs</b> <b>30.0%</b>	<ul style="list-style-type: none"> <li>● Labs</li> <li>● Includes guided inquiry, open ended research, and content-based skill labs.</li> </ul>
<b>Assignments</b> <b>10%</b>	<ul style="list-style-type: none"> <li>● A variety of assignments that are given to learn the material, students need to complete the assignments in order to learn the material.</li> <li>● All assignments must be done in order to retest</li> <li>● A list of some examples, but not limited to               <ul style="list-style-type: none"> <li>○ Google Classroom assignments (not including labs)</li> <li>○ Modeling activities</li> <li>○ Practice games</li> <li>○ Videos</li> <li>○ Surveys</li> </ul> </li> </ul>
<b>How to be successful in Chemistry</b>	<ul style="list-style-type: none"> <li>● Actively seek understanding by watching instructional videos and/or reading the textbook</li> <li>● Take notes that help you synthesize and identify key information</li> <li>● Fully engage and participate in activities with a positive mindset</li> <li>● Communicate concerns through email or individually before or after class</li> <li>● Reach out to me with suggestions</li> </ul>
<b>Attendance Policy</b>	<p>Attendance is a key part of success. If you are gone, it is your responsibility to check google classroom for assignments and communicate with me. If you miss a major lab or assessment, you will need to make arrangements to come in during PACK time, before or after school, or complete an alternative virtual lab.</p>

